



How do polymers degrade in Antarctica?

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&

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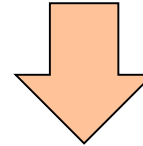
Environmental testing - Weathering

Polymer weathering:

Effect of environmental impacts on polymer properties
Verification of ability to withstand intended application
Almost realized in localities with high intensity of solar radiation

- Solar radiation
- Temperature
- Humidity
- Rain
- Mechanical stress
- Air-pollutions

*All varying
in intensity and time*



Pilot localities used for polymer testing (yearly doses):

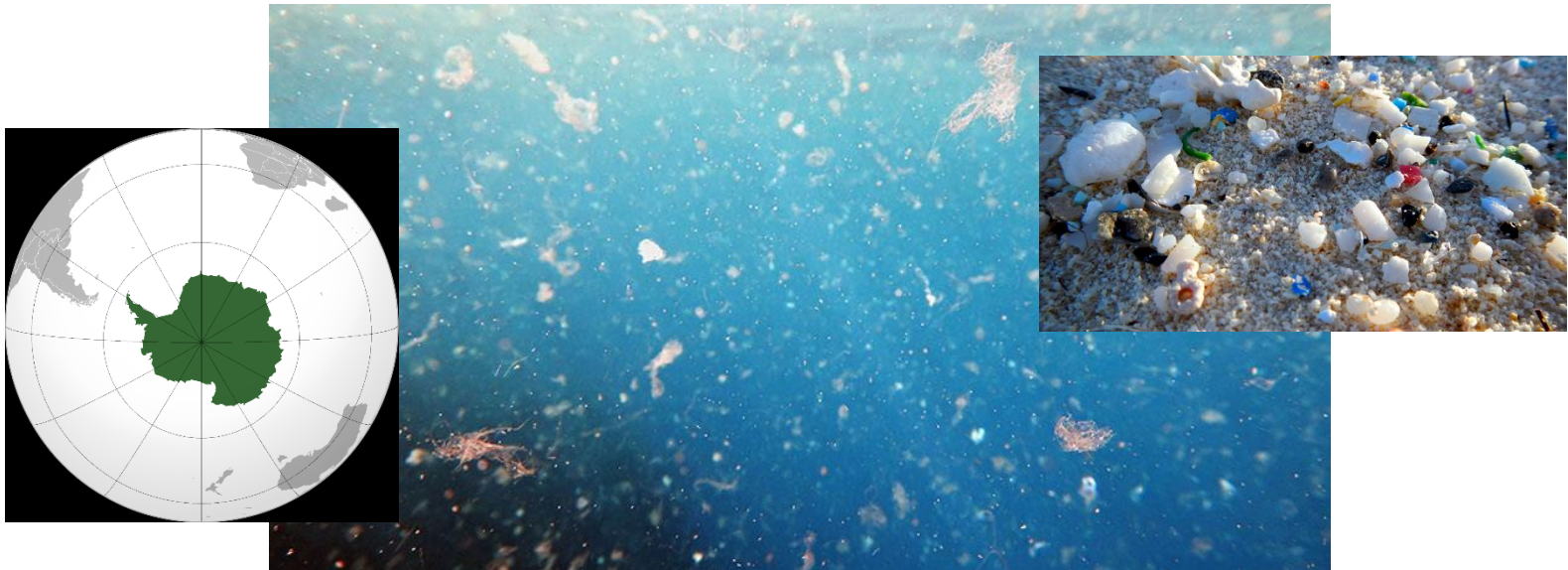
Florida	140 kLy	(marine, wet)
Arizona	180 kLy	(continental, dry)
Bandol, F (south EU)	120 kLy	(marine, wet)
Basel, Swiss (middle EU)	90 kLy	(subcontinental, mild)

Weathering tests provide the most reliable information on polymer durability under the real climatic conditions of application.

In general, weathering of polymers may be carried out anywhere on the planet !

Why Antarctica?

- High social attention to marine pollutions caused by plastics in general
- Commodity polymers particles found as microplastics by the coast of Antarctica



Questions:

What happens with polymer exposed to Antarctic climate ?

How long will it remain there ?

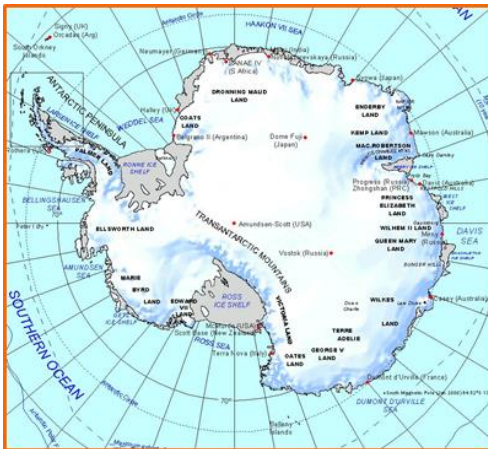
Why Antarctica?

Presently, more than **80** scientific stations located in Antarctica and more than **3000** people permanently living there as staff of the stations

Polymers are used as parts of buildings, vehicles, instruments, packaging and garments.

Specific environmental conditions:

- Extremely low temperatures
- Short diurnal period
- Long polar nights
- Short summer period



POLYMER FRIENDLY?

But:

- Existence of ozone depletion (ozone hole)
- High and fast diurnal changes of temperature
- Strong winds drifting ice crystals

Antarctic climate - how is the overall effect on polymer and its stability?

Up to 2019, no information on polymer environmental degradation in Antarctica was published in scientific literature (WOS, Scopus)

J.G.Mendel Polar Scientific Station

Located on James Ross Island in Antarctica (S 63°49', W 57°53')

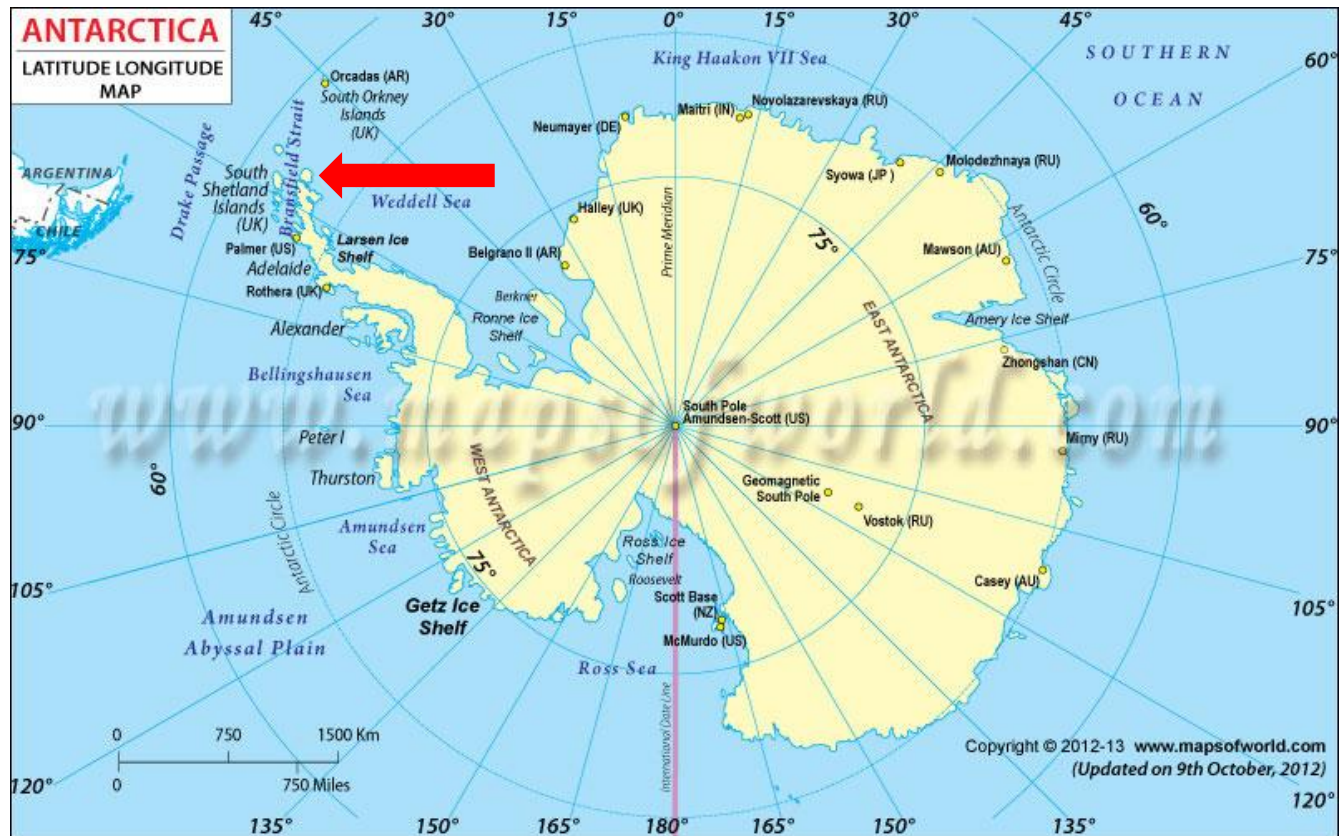
Built and operated by Masaryk University Brno (Czech Republic) since 2005



J.G.Mendel Polar Scientific Station

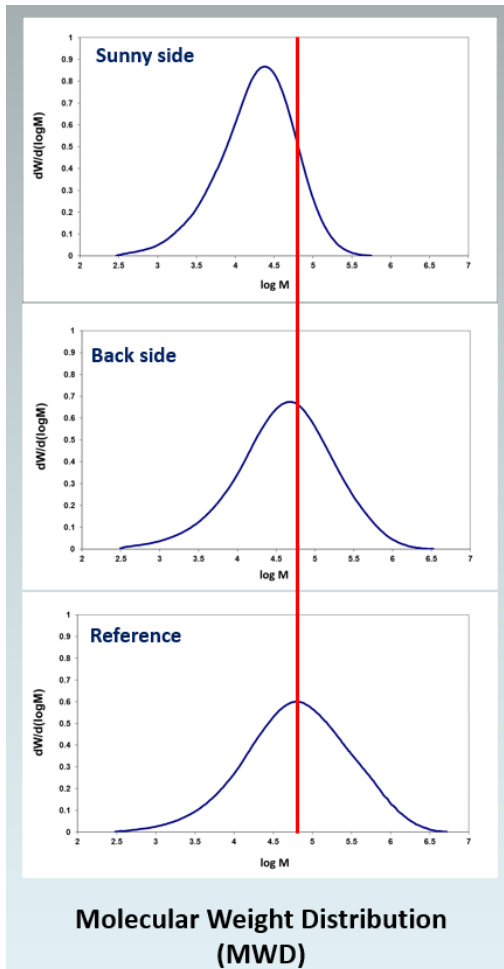
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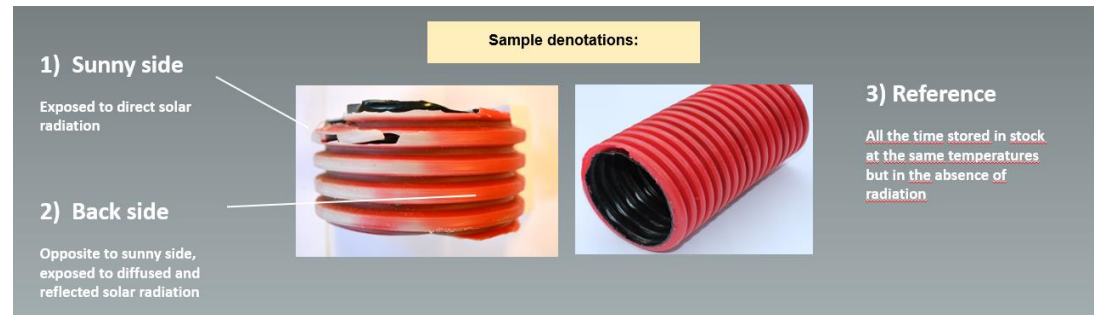
Environmental degradation in Antarctica

First observations (2014):



High density polyethylene (HDPE) - extruded, corrugated pipe

Diameter: 110 mm
Wall thickness: 0.65 ± 0.05 mm
Stabilization: Processing, no UV



The pipe was exposed **9 years outdoors** in Antarctica on James Ross Island (S 63°49', W 57°53')

Serving as mechanical protection of electrical cabling installed between J.G. Mendel Station and its remote scientific facilities

Environmental degradation in Antarctica

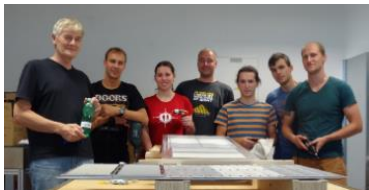
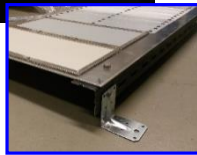
Long-term weathering experiment 2015-2020:

Materials: Polypropylene (PP) – homopolymer (random copolymer, impact copolymer)

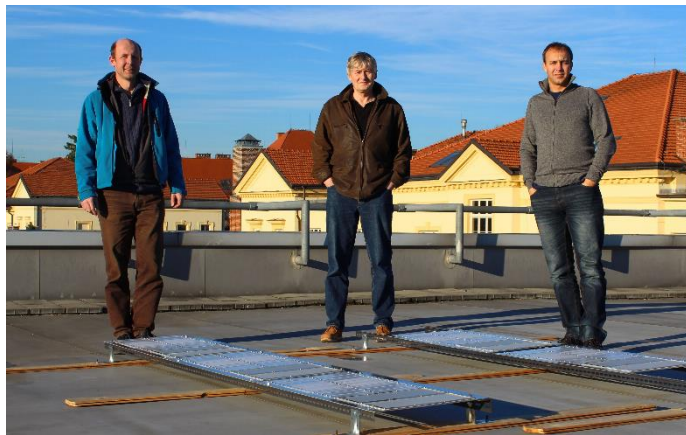
Exposure sites:
a) James Ross Island (Antarctica)
b) Brno, Czech Republic (Central Europe)

Time-span: 5 years or + / sampling once a year

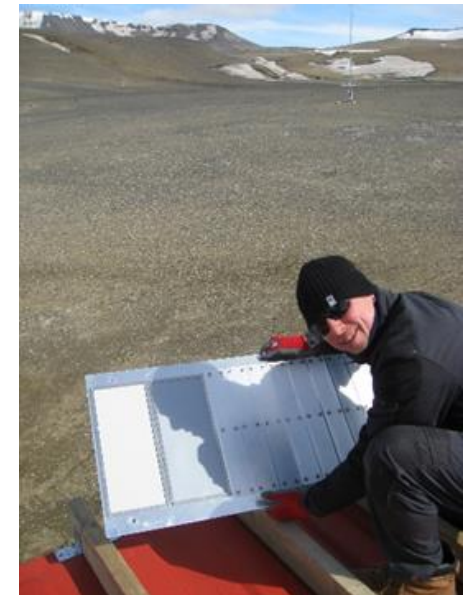
Aluminium carrier racks



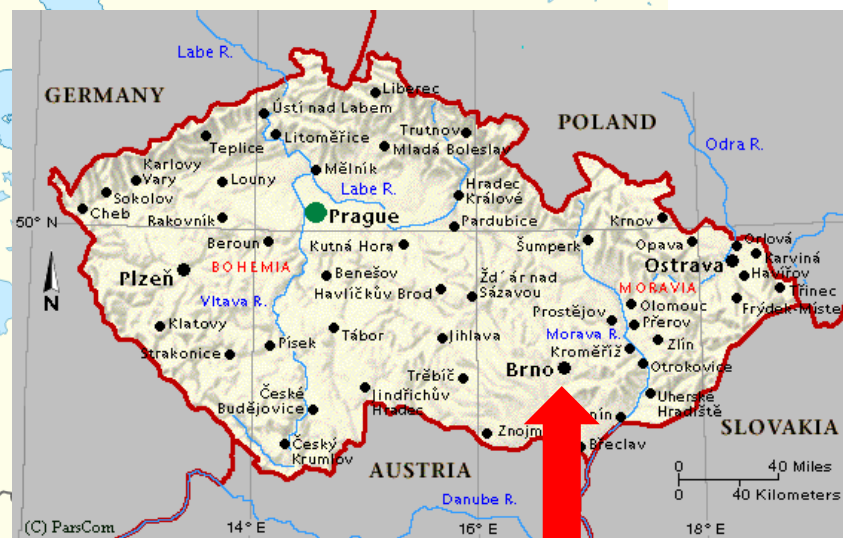
Exposure site in Brno, CZ (Central EU)



Exposure site
James Ross Island (Antarctica)



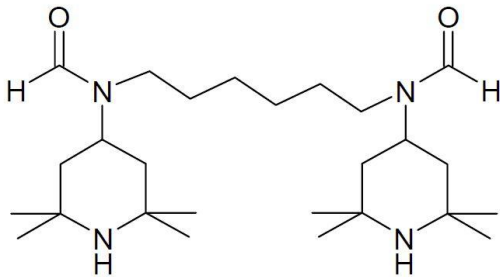
EUROPE



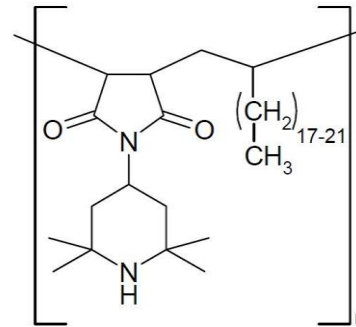
BRNO
Reference exposure site

Environmental degradation in Antarctica

- PP-homopolymer:** Mosten (Unipetrol RPA), isotactic homopolymer, gas-phase process
MFI=3 dg/min. (230°C/2.16 kg; ISO 1133)
- Sample geometry:** Extruded sheets 0.5 mm thick (chill-roll technology)
- Additivation (tailor-made):** *Base:* 0.15% Irganox B225, 0.02% Ca-stearate SP, 0.02% Hycite 713
Light stabilization: a) *none*
b) 0.1% HALS-1
c) 0.1% HALS-2



HALS-1 (Uvinul 4050H)
UV stabilizer, low-molecular type
(CAS No. 124172-53-8; Mol. weight 450 g/mol)



HALS-2 (Uvinul 5050H)
UV stabilizer, oligomeric type
(CAS No. 152261-33-1; Mol. weight 3000-4000 g/mol)

Environmental conditions

Climatic conditions were continuously recorded at both exposure sites

Antarctica (JGM station) – overview of environmental data in the years 2015–2017.

Period	Months	T _{min} (°C)	T _{max} (°C)	T _{avg} (°C)	UV-A (MJ/m ²)	UV-B (MJ/m ²)	TUVR (MJ/m ²)	Global (MJ/m ²)	TOC (DU)
9.1.2015–6.2.2016	13	–11.0	–3.1	–7.2	297.8	5.5	303.3	4272.6	280.8
7.2.2016–21.1.2017	11	–8.4	–1.3	–5.1	214.1	3.4	217.5	3200.9	283.6
22.1.2017–19.1.2018	12	–11.1	–3.2	–7.3	268.6	4.4	273.0	3620.8	285.9
<i>Total:</i>					780.5	13.3	793.8	11 094.3	283.4 ^a

Brno (Czech Republic) – overview of environmental data in the years 2015–2017.

Period	Months	T _{min} (°C)	T _{max} (°C)	T _{avg} (°C)	UV-A (MJ/m ²)	UV-B (MJ/m ²)	TUVR (MJ/m ²)	Global (MJ/m ²)	TOC (DU)
13.1.2015–9.2.2016	13	6.1	14.9	10.5	287.5	4.6	292.2	4 305.7	330.4
10.2.2016–13.1.2017	11	6.5	15.7	11.1	255.8	4.6	260.3	4 013.2	322.3
14.1.2017–12.1.2018	12	6.3	15.5	10.9	273.7	5.0	278.7	4 360.2	320.4
<i>Total:</i>					817.0	14.2	831.2	12 679.1	324.4 ^a

T_{min} - average of daily minimum temperatures.

T_{max} - average of daily maximum temperatures.

T_{avg} - average of daily average temperatures.

^a average value.

Incident solar radiation ranges:

UV-B	280-315 nm
UV-A	315-400 nm
Global	280-2800 nm
TUVR =	UV-A + UV-B

Radiation energy ratio:	Antarctica (%)	Brno (%)
UV-A / Global	7.0	6.4
UV-B / Global	0.12	0.11
TUVR / Global	7.2	6.6
UV-B / UV-A	1.7	1.73

Based on 3-year cumulative data

Environmental degradation in Antarctica

Carbonyl Index (CI)

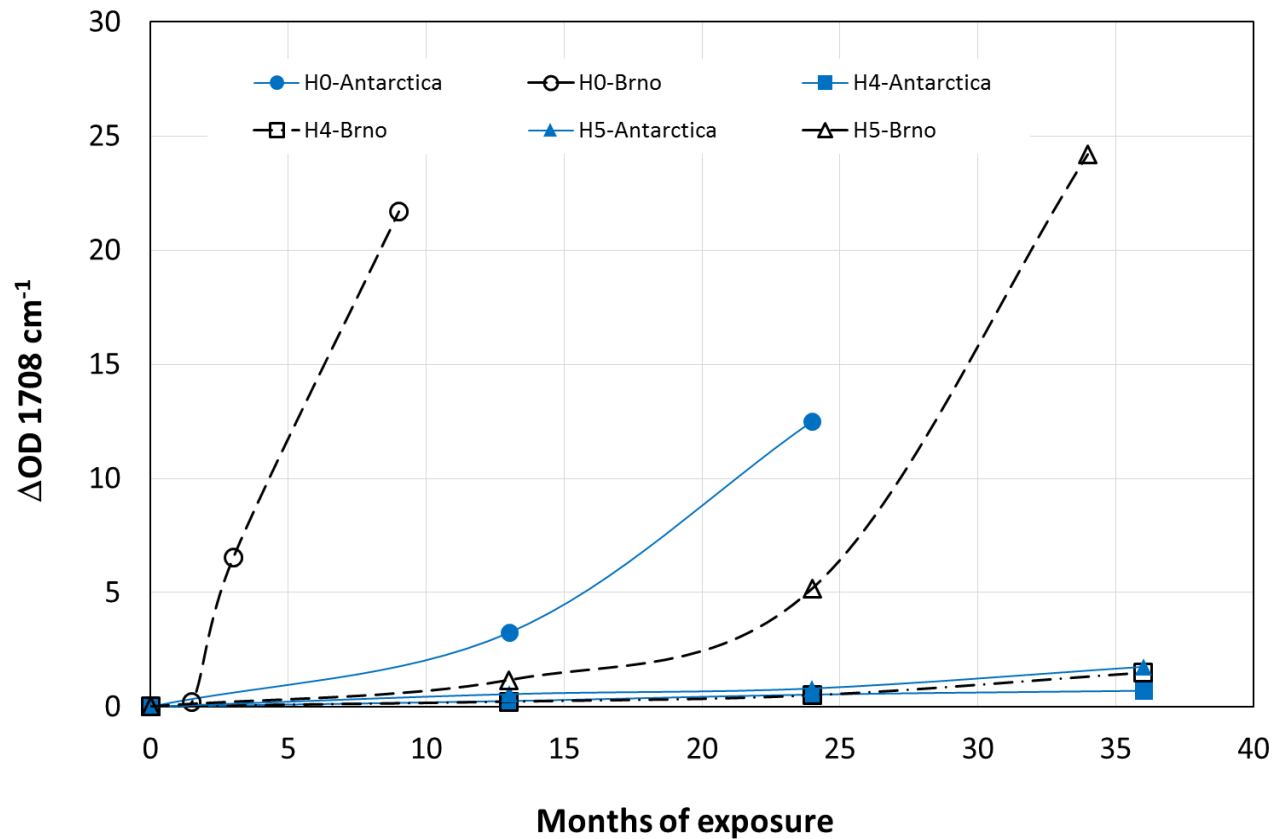
$$CI = A_{1708} / A_{1892}$$

A_{1708}, A_{1892} absorbances of carbonyls and reference band, respectively

Optical density (OD)

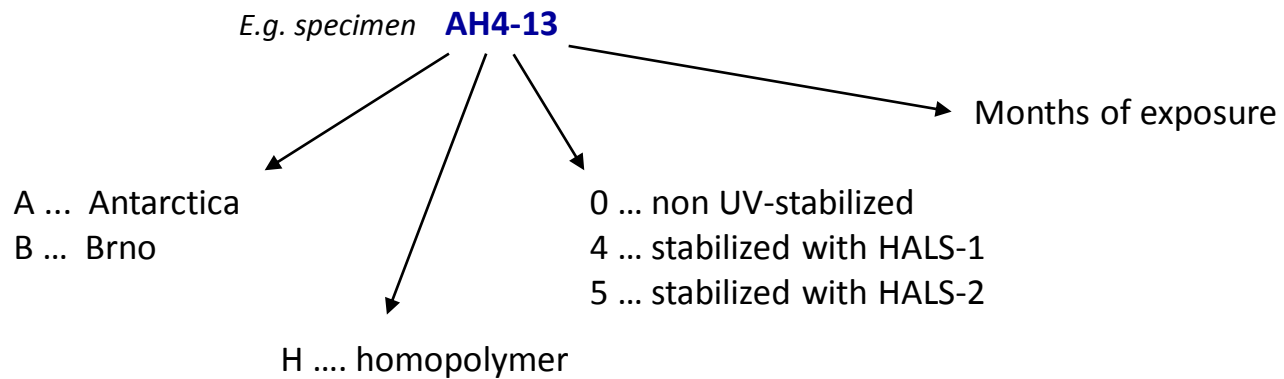
$$\Delta OD = CI_t - CI_0$$

CI_t, CI_0 carbonyl indices at time of sampling and before exposure, respectively



Environmental degradation in Antarctica

Denotations of test specimens:



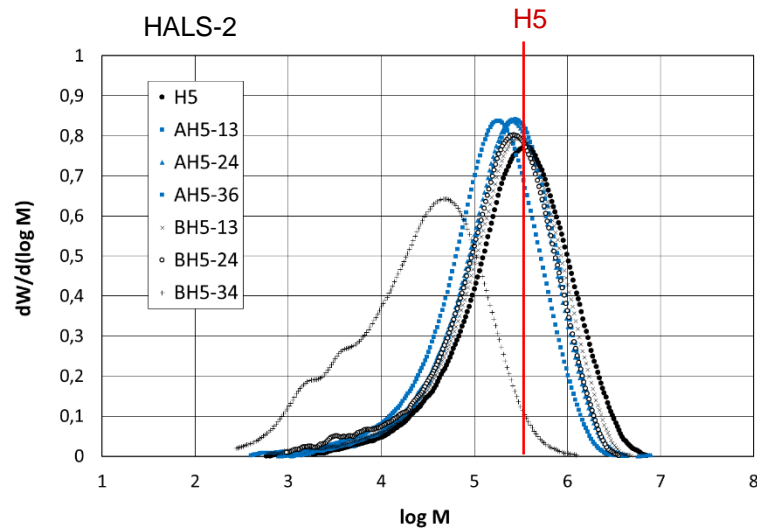
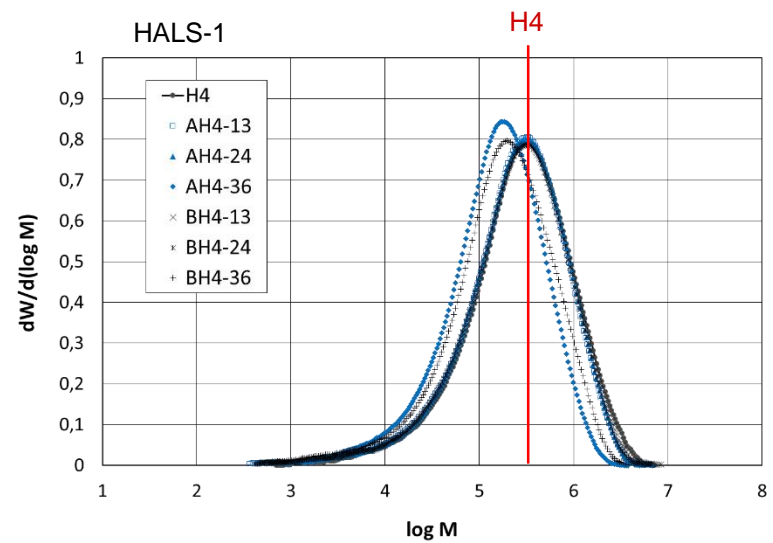
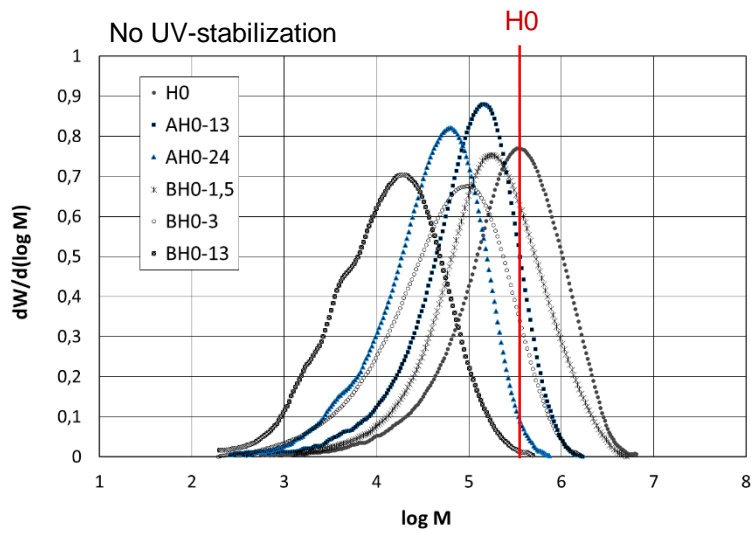
Denotations H0, H4 and H5 stand for the non-exposed polymers.

Molecular weight (GPC)

Average molecular weights in thousands g/mol and polydispersity

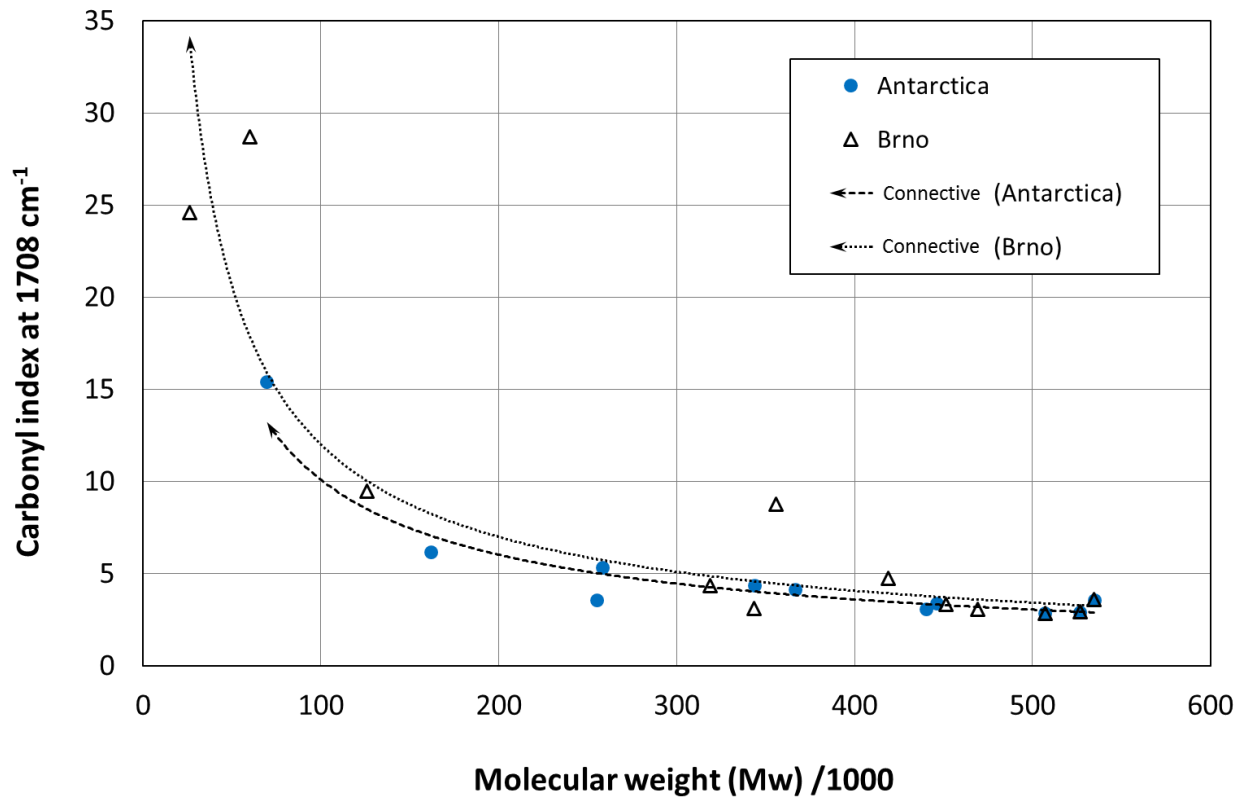
Sample	M_w	M_n	M_w/M_n
H0	527	75	7.0
H4	507	78	6.5
H5	535	75	7.1
AH0-13	162	32	5.1
AH0-24	70	14	5.0
BH0-1.5	344	60	5.8
BH0-3	126	15	8.3
BH0-9	26	5.8	4.5
BH0-13	28	5.3	5.3
AH4-13	440	55	8.0
AH4-24	446	55	8.2
AH4-36	255	44	5.9
BH4-13	469	56	8.4
BH4-24	451	51	8.9
BH4-36	319	42	7.6
AH5-13	367	72	5.1
AH5-24	344	68	5.1
AH5-36	258	41	6.3
BH5-13	419	51	8.2
BH5-24	356	47	7.6
BH5-34	60	6.6	9.1

Molecular weight (GPC)



Environmental degradation in Antarctica

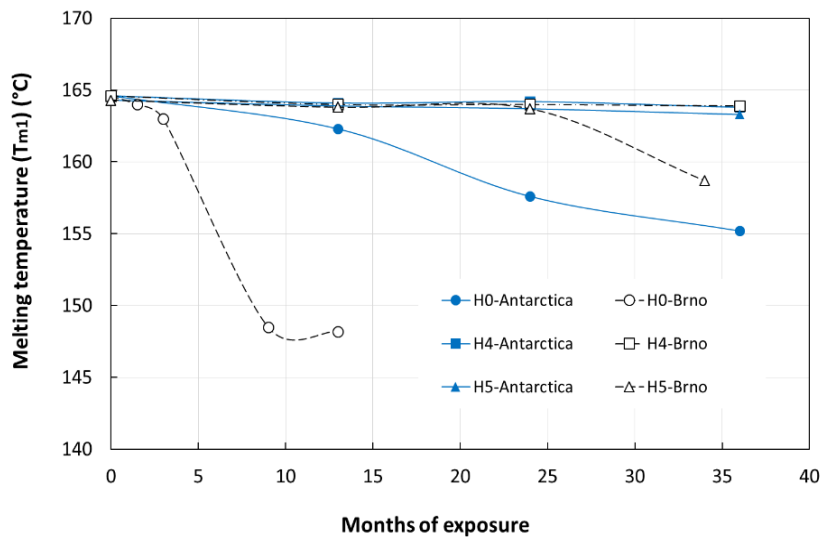
PP-homo: Intensive carbonyls formation mostly induced by dramatic drop in weight average molecular weight at both localities



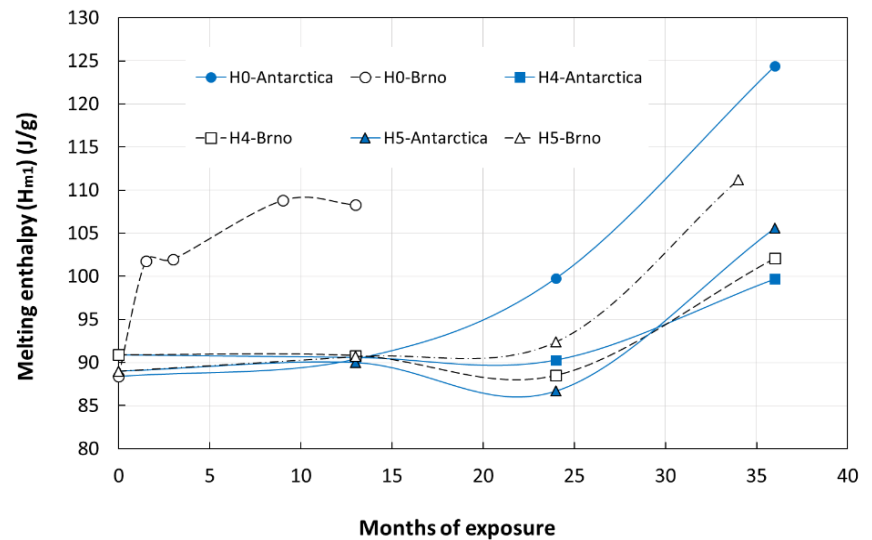
Thermal analyses - DSC

PP-homopolymer:

Melting temperature



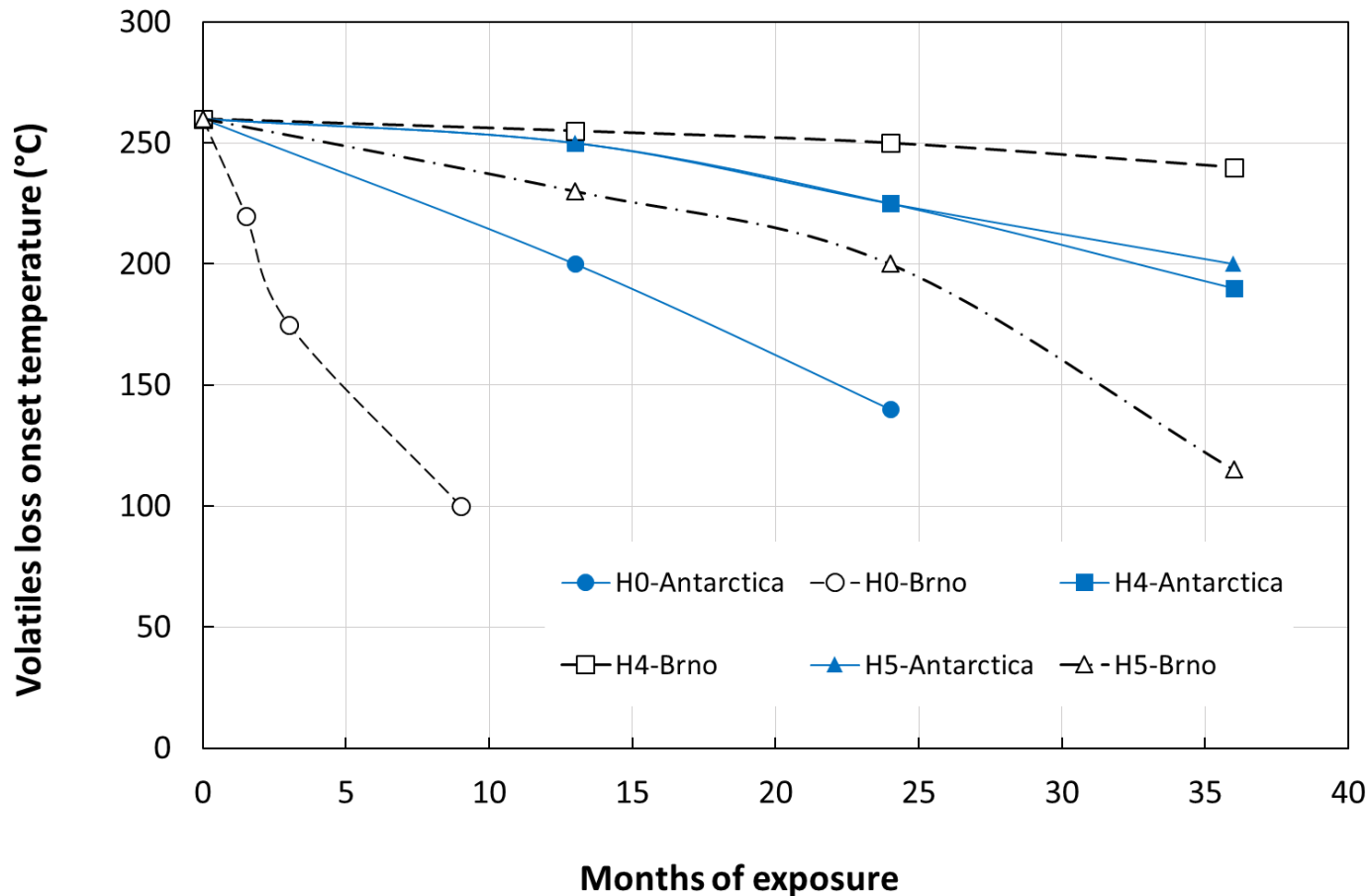
Crystallinity



- Proceeding degradation decreased the molecular weight – drop in melting temperature (T_{m1})
- Shorter macromolecules exhibiting better mobility lead to increased crystallinity ($\sim H_{m1}$)

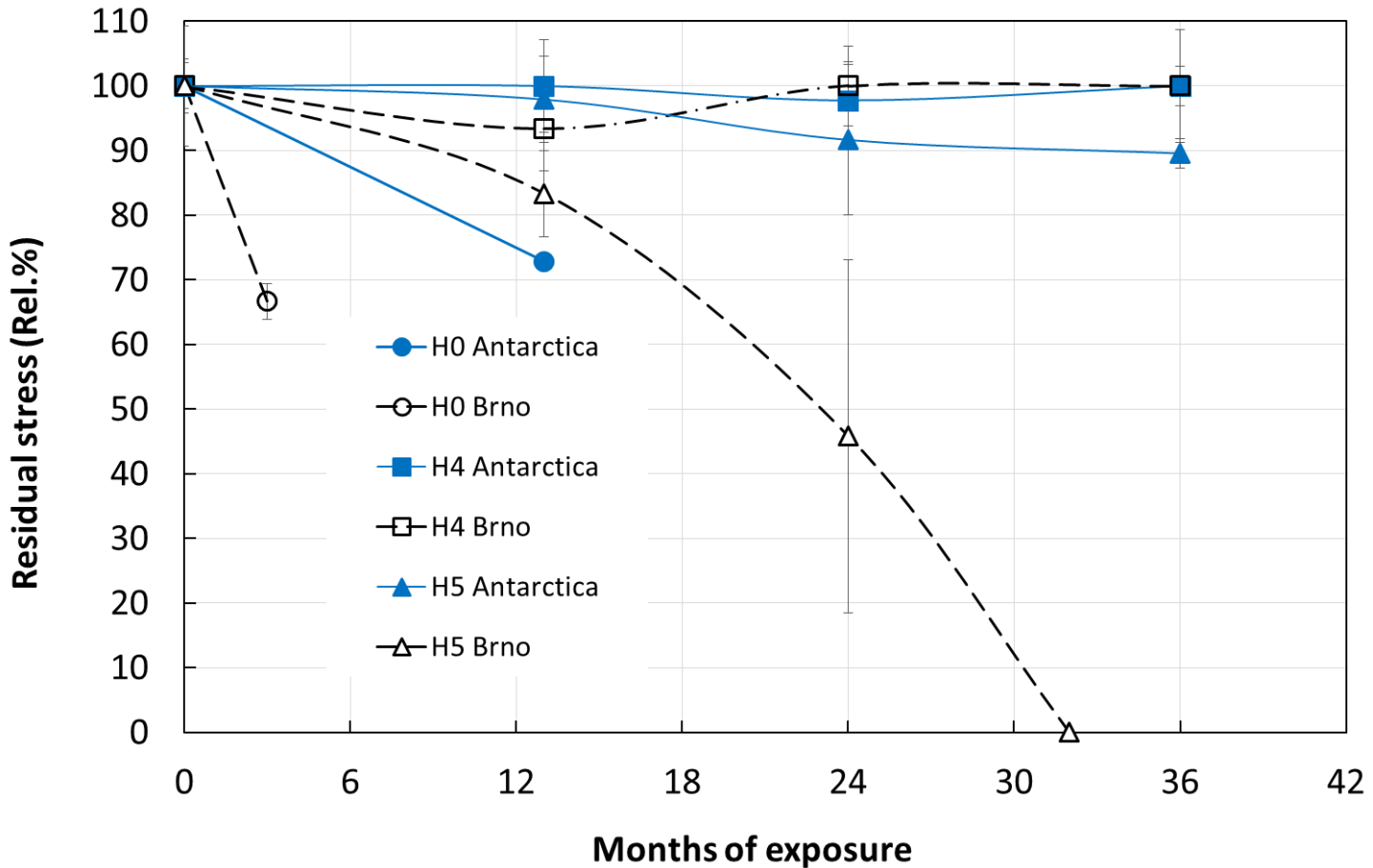
Thermogravimetric analysis (TGA)

Volatiles loss onset (VLO) – defined as temperature at which formation of volatiles exceeds 0.1 wt.%



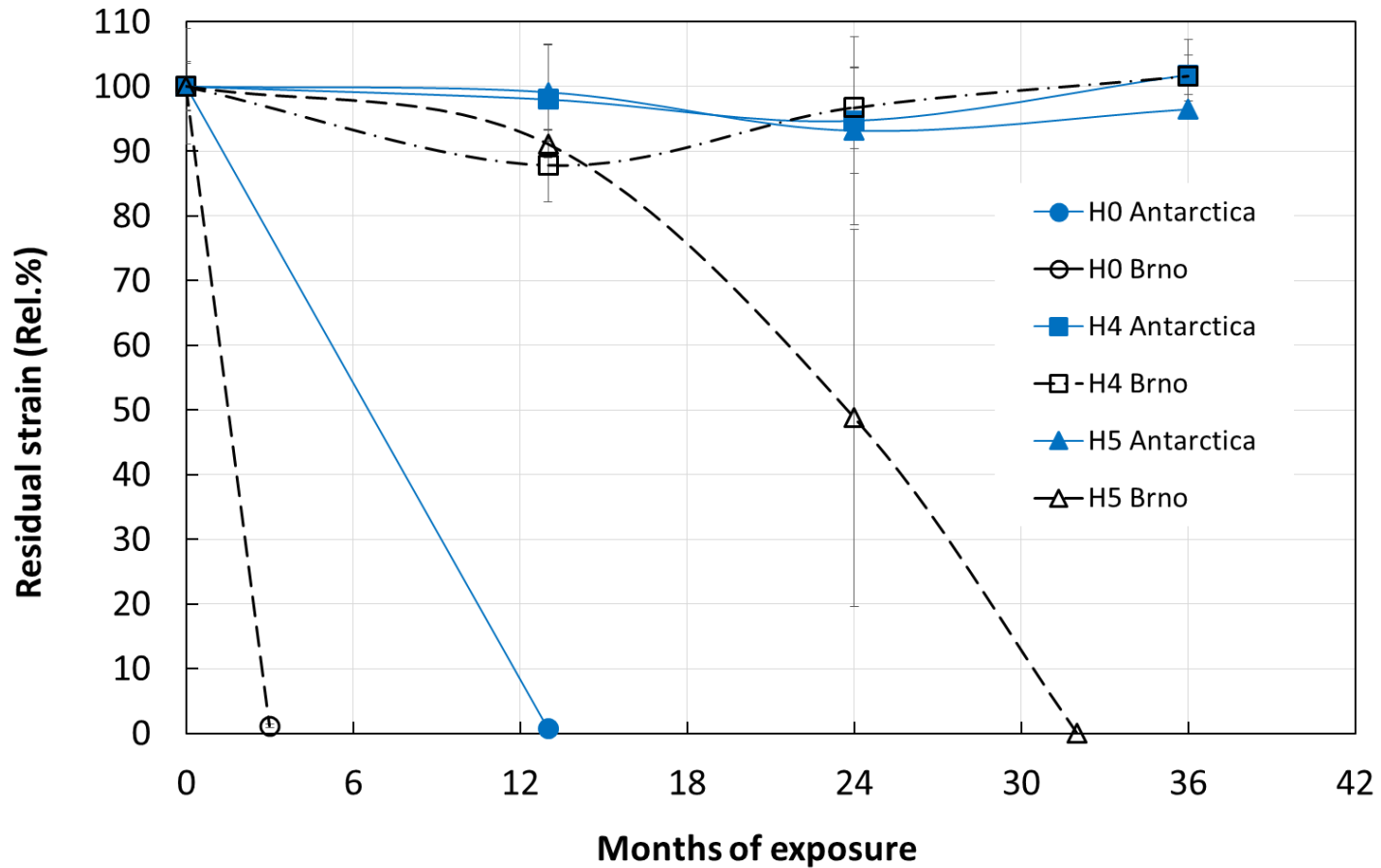
Mechanical properties – tensile test

PP-homopolymer: 0.5 m film, dog-bone tensile test specimens, grip distance 50 mm, crosshead speed 50 mm/min.



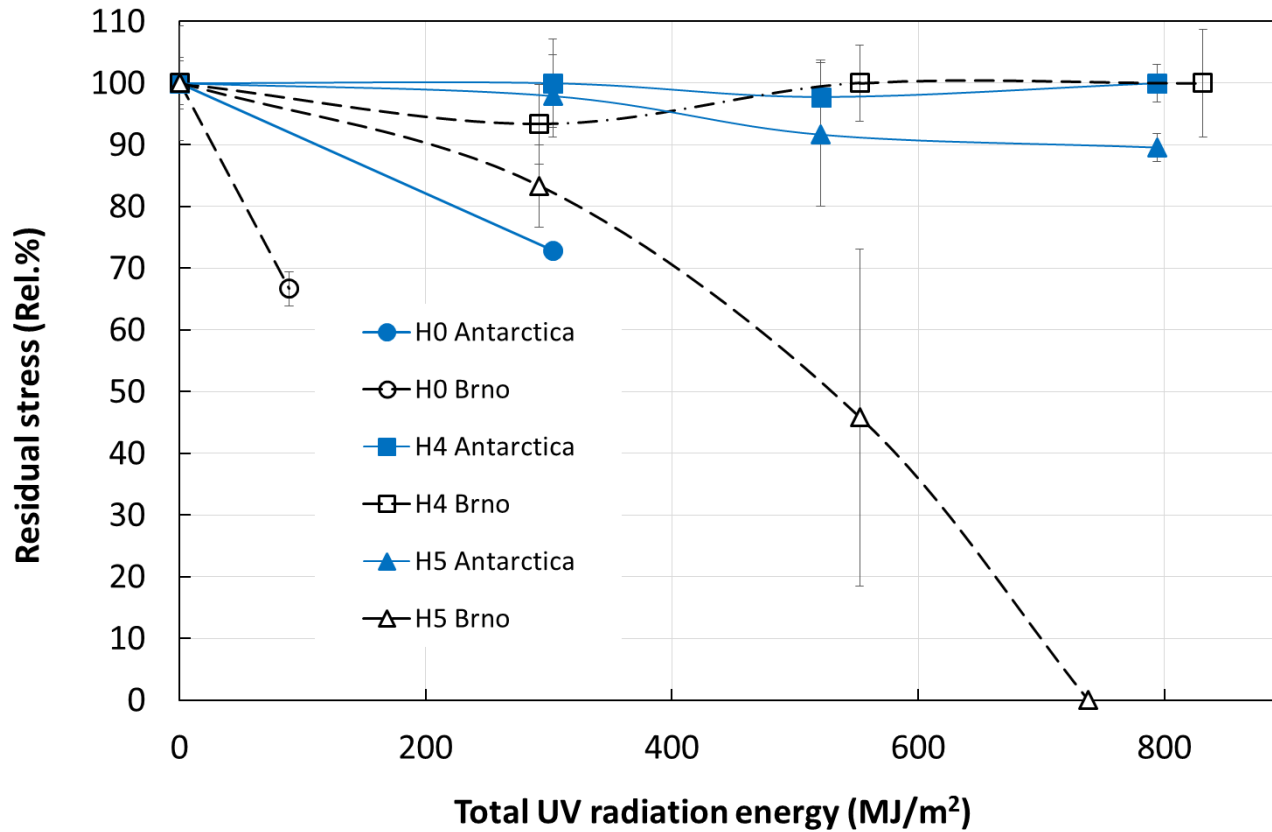
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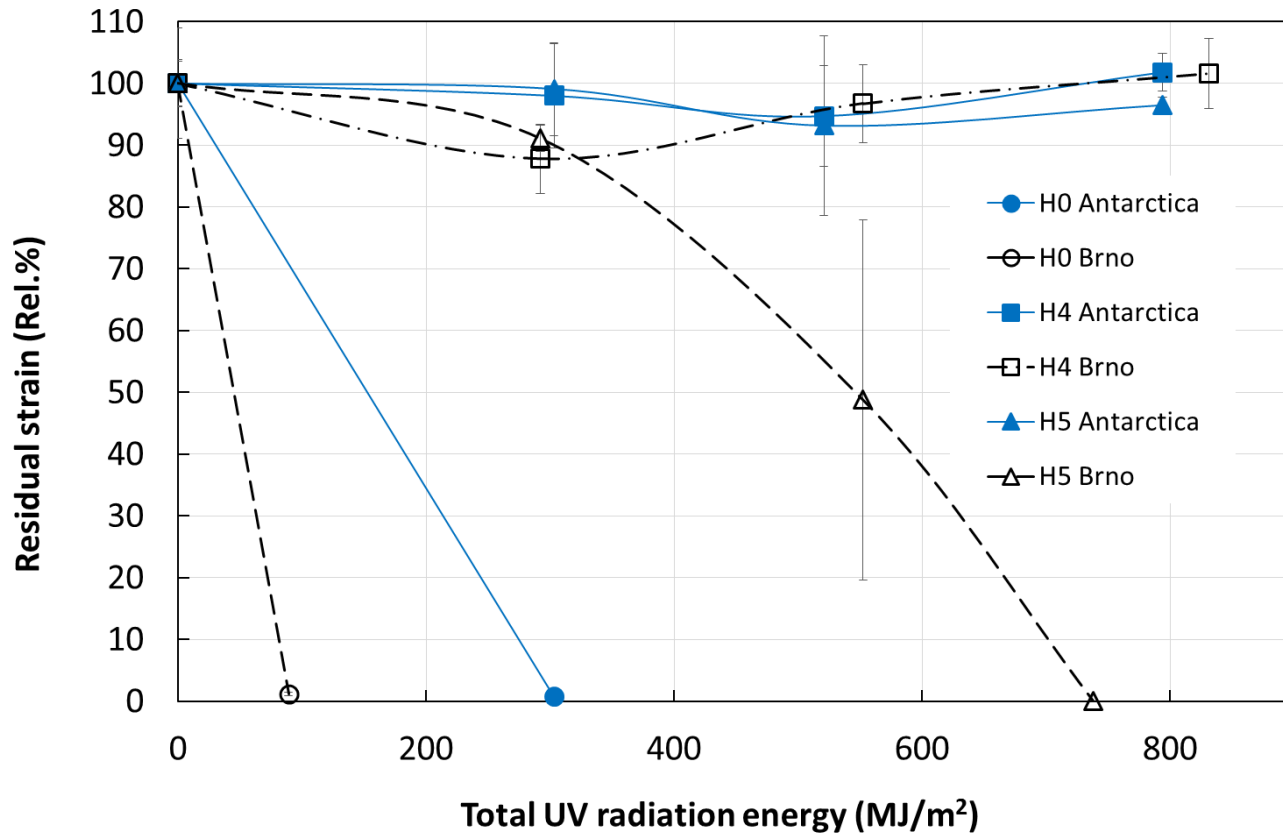
Mechanical properties vs. radiant energy

PP-homopolymer: 0.5 m film, dog-bone tensile test specimens, grip distance 50 mm, crosshead speed 50 mm/min.



Mechanical properties vs. radiant energy

PP-homopolymer: 0.5 m film, dog-bone tensile test specimens, grip distance 50 mm, crosshead speed 50 mm/min.



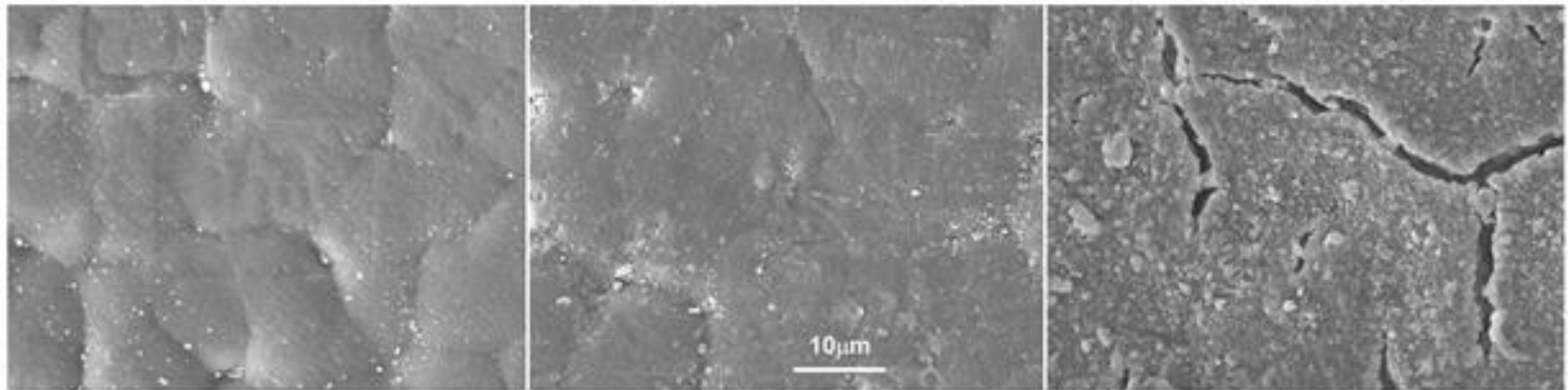
Surface morphology – SEM imaging / 13 months

Non-UV-stabilized polymer – surface deterioration after 13 months of outdoor exposure

Antarctica – minimum visually observable changes on the surface

Brno – total degradation of both surface/bulk

Magnification 10 000 x



H0

AH0-13

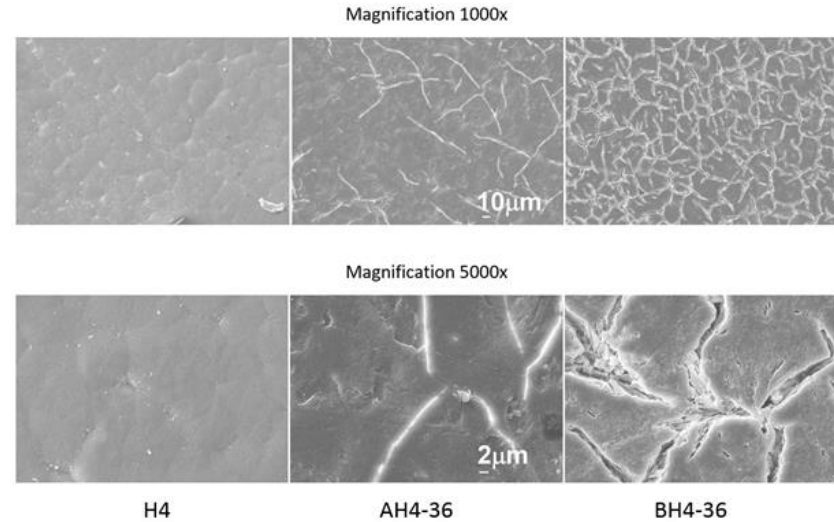
BH0-13

Surface morphology - SEM imaging / 36 months

HALS-1

Mostly surface deterioration at both localities

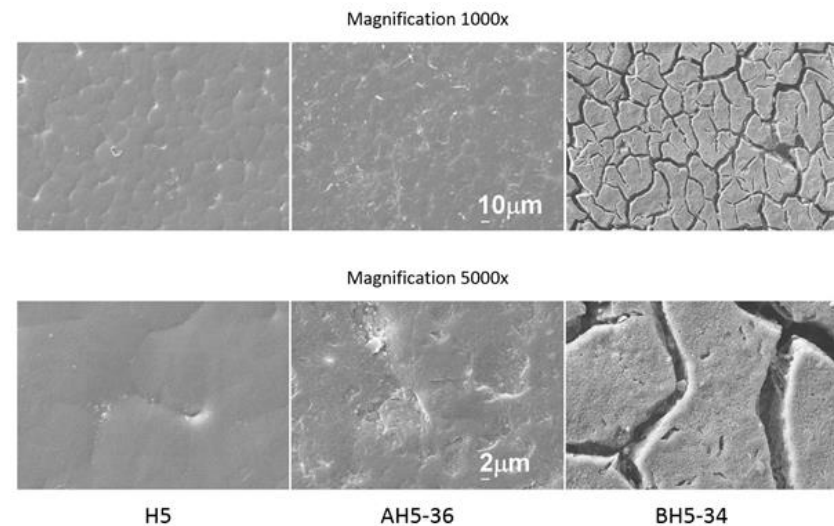
Degradation allowed by extraction of low-molecular HALS from the surface layer of polymer



HALS-2

Slight surface degradation in Antarctica
Bulk degradation in Brno

Low stabilization efficiency of oligomeric HALS at higher ambient temperatures



Kinetics

Estimation of polymer service-life based on mechanical properties:

	Months	Property dropped by 50%
H0-Antarctica	6	strain
H0-Brno	1.5	strain
H4-Antarctica	> 36	n.a.
H4-Brno	> 36	n.a.
H5-Antarctica	> 36	n.a.
H5-Brno	23	stress

4 x

If weathering was controlled only by temperature:

$$\ln k_2/k_1 = -E_a/R (1/T_2 - 1/T_1)$$

$$k_2/k_1 = 2.8$$

k_2, k_1 rate constants in Brno and Antarctica, respectively

E_a activation energy (37.9 kJ/mol)

R gas constant (8.314 J/°K/mol)

T_2, T_1 absolute temperatures in Brno and Antarctica, respectively

OVERALL: In Antarctica degradation is mostly controlled by solar radiation, while temperature plays only marginal role
In Brno degradation is controlled both by solar radiation and temperature

ANTARCTICA

36 months



BRNO

36 months



Airborne pollutions demonstrated on roofing materials tested together with PP

Conclusions:

- Antarctica's environmental conditions seem to be at first sight polymer friendly but even here synthetic polymers degrade
- Antarctic climatic conditions are characterized by very low ambient temperatures and a higher portion of TUVB in the global radiation as a consequence of ozone hole. Both compared to Central Europe (CE)
- Environmental degradation of PP-homopolymer was found to be faster in CE, the difference, however, was lower than we expected before starting this experiment
- Higher degradation rate in CE may be accounted for the higher absolute yearly dose of radiant energy and higher ambient temperatures. Airborne pollutions in CE may also have contributed
- Efficient UV stabilization of polypropylene against Antarctic climatic conditions with both low-molecular and oligomeric HALS stabilizers is possible

**Even in Antarctica synthetic polymers degrade,
compared to Central Europe at a slower rate but definitely not negligibly !**

Thank you for your attention !

